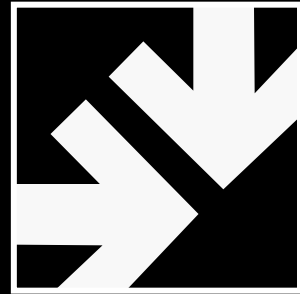
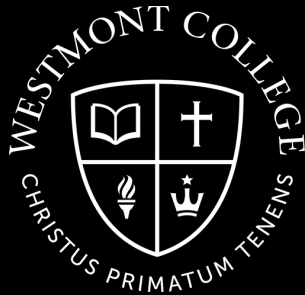


“Does the Second Law of Thermodynamics Contradict the Theory of Evolution?”

Michael A. Everest



Your Background

Your Background

- Secular College/University

Your Background

- Secular College/University
- Very Christian College/University

Your Background

- Secular College/University
- Very Christian College/University
- Nominal Christian College/University

Your Background

- Secular College/University
- Very Christian College/University
- Nominal Christian College/University
- Scientist at Industry/Government Lab

Your Background

- Secular College/University
- Very Christian College/University
- Nominal Christian College/University
- Scientist at Industry/Government Lab
- Non-Scientist

Your Background

Your Background

- Chemistry

Your Background

- Chemistry
- Physics

Your Background

- Chemistry
- Physics
- Geology

Your Background

- Chemistry
- Physics
- Geology
- Biology

Your Background

- Chemistry
- Physics
- Geology
- Biology
- Mind/Social Science

Your Background

- Chemistry
- Physics
- Geology
- Biology
- Mind/Social Science
- Engineering

Your Background

- Chemistry
- Physics
- Geology
- Biology
- Mind/Social Science
- Engineering
- Other Science

Your Background

- Chemistry
- Physics
- Geology
- Biology
- Mind/Social Science
- Engineering
- Other Science
- Humanities

Your Background

- Chemistry
- Physics
- Geology
- Biology
- Mind/Social Science
- Engineering
- Other Science
- Humanities
- Theology/Ministry

My Background

My Background

- Background: chemical reaction dynamics

My Background

- Background: chemical reaction dynamics
- Currently: interfacial chemistry

My Background

- Background: chemical reaction dynamics
- Currently: interfacial chemistry
- Teach undergraduate physical chemistry

My Background

- Background: chemical reaction dynamics
- Currently: interfacial chemistry
- Teach undergraduate physical chemistry
- *Not* a specialist in thermodynamics

My Background

- Background: chemical reaction dynamics
- Currently: interfacial chemistry
- Teach undergraduate physical chemistry
- *Not* a specialist in thermodynamics
- Primarily an *Educator*

My Context



My Context



- Westmont College

My Context



- Westmont College
- 1200 students on campus

My Context



- Westmont College
- 1200 students on campus
- 12 Chemistry majors/yr

My Context



- Westmont College
- 1200 students on campus
- 12 Chemistry majors/yr
- Top 100 Liberal Arts

My Context



- Westmont College
- 1200 students on campus
- 12 Chemistry majors/yr
- Top 100 Liberal Arts
- Undergraduate only

My Context



- Westmont College
- 1200 students on campus
- 12 Chemistry majors/yr
- Top 100 Liberal Arts
- Undergraduate only
- Mostly Christian students

My Context



- Westmont College
- 1200 students on campus
- 12 Chemistry majors/yr
- Top 100 Liberal Arts
- Undergraduate only
- Mostly Christian students
- 40% YEC background

Some Science: What is the Second Law?

Some Science: What is the Second Law?

It is not possible to move heat from a cold place to a hot place without something else happening.

Some Science: What is the Second Law?

It is not possible to move heat from a cold place to a hot place without something else happening.

It is not possible to convert 100 Joules of heat into 100 Joules of work.

Some Science: What is the Second Law?

It is not possible to move heat from a cold place to a hot place without something else happening.

It is not possible to convert 100 Joules of heat into 100 Joules of work.

Etc.

More Science: What is Entropy?

More Science: What is Entropy?

- Two definitions

More Science: What is Entropy?

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_{\text{B}} \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_{\text{B}} \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_B \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.



More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_{\text{B}} \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_B \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.



More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_{\text{B}} \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.

More Science: What is Entropy?

$$dS = \frac{dq_{\text{rev}}}{T}$$

$$S = k_{\text{B}} \ln \Omega$$

- Two definitions
- “Thermodynamic”
 - How much heat added at what temperature?
- “Statistical”
 - How many ways can it happen?
- Only one way to get all heads. Lots of ways to get half heads and half tails.
- Occasionally “disorder” or “randomness”

More Science: What is the Second Law?

More Science: What is the Second Law?

The entropy of an isolated system can only stay the same or increase.

More Science: What is the Second Law?

The entropy of an isolated system can only stay the same or increase.

The entropy of the universe tends toward a maximum.

More Science: What is the Second Law?

The entropy of an isolated system can only stay the same or increase.

The entropy of the universe tends toward a maximum.



System

More Science: What is the Second Law?

The entropy of an isolated system can only stay the same or increase.

The entropy of the universe tends toward a maximum.

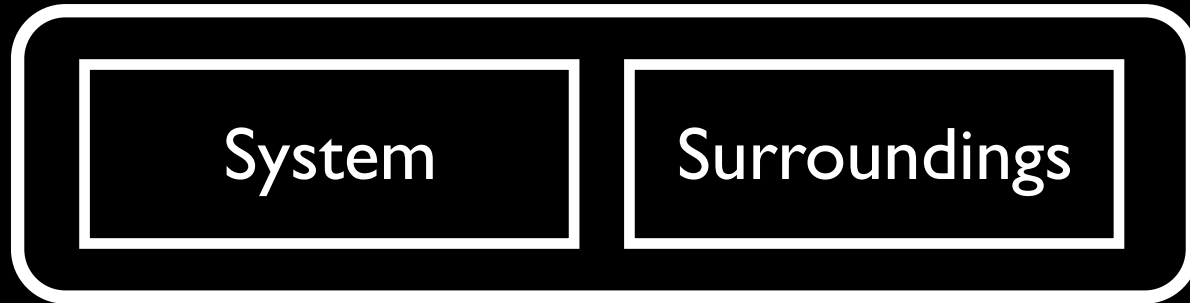
System

Surroundings

More Science: What is the Second Law?

The entropy of an isolated system can only stay the same or increase.

The entropy of the universe tends toward a maximum.



Last Science: What is Evolution?

Last Science: What is Evolution?

Earth with *no* biosphere →

Last Science: What is Evolution?

Earth with *no* biosphere → Earth *with* a biosphere

The Assignment: Intro

The Assignment: Intro

- Imagine that ...

The Assignment: Intro

- Imagine that ...
- They want to know...

The Assignment: Intro

- Imagine that ...
- They want to know...
- They have asked you for...

The Assignment: Details

The Assignment: Details

- ...at least two possible approaches

The Assignment: Details

- ...at least two possible approaches
- Your recommendation

The Assignment: Details

- ...at least two possible approaches
- Your recommendation
- At least five sources.

The Assignment: My Goals

The Assignment: My Goals

- Encourage faith/science reflection

The Assignment: My Goals

- Encourage faith/science reflection
 - “contribution” vs. “integration”

The Assignment: My Goals

- Encourage faith/science reflection
 - “contribution” vs. “integration”
- Encourage service to the church

The Assignment: My Goals

- Encourage faith/science reflection
 - “contribution” vs. “integration”
- Encourage service to the church
- Encourage irenic thinking about a “controversial” issue

The Assignment: My Goals

- Encourage faith/science reflection
 - “contribution” vs. “integration”
- Encourage service to the church
- Encourage irenic thinking about a “controversial” issue
- Direct them to the literature

Answers: The Obvious

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

Answers: The Obvious

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

- No.

Answers: The Obvious

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- No.
- The Sun is an essentially limitless source of energy, so the Earth is not an isolated system.

Answers: The Obvious

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

Answers: The Obvious

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

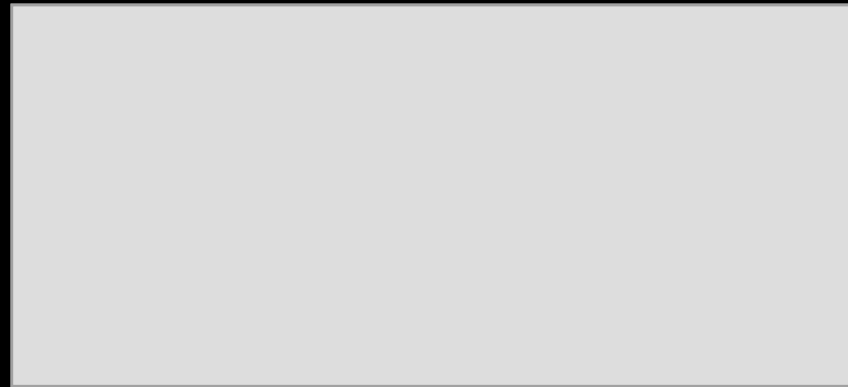
CO₂, H₂O



Answers: The Obvious

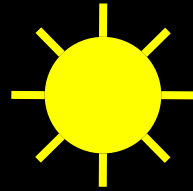
Does the Second Law of Thermodynamics contradict the Theory of Evolution?

$\text{CO}_2, \text{H}_2\text{O}$



Answers: The Obvious

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

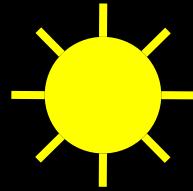


$\text{CO}_2, \text{H}_2\text{O}$

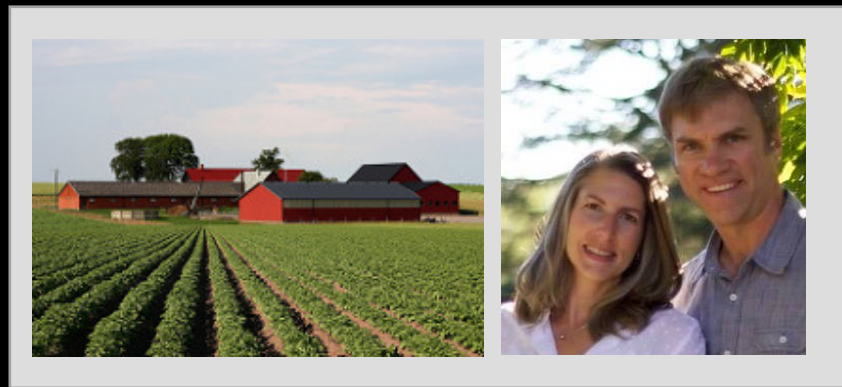


Answers: The Obvious

Does the Second Law of Thermodynamics contradict the Theory of Evolution?



$\text{CO}_2, \text{H}_2\text{O}$



Answers: What Students Find

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

Answers: What Students Find

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- A few books and websites say, “Yes”

Answers: What Students Find

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- A few books and websites say, “Yes”
- Lots of physics articles say, “No”

Answers: What Students Find

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- A few books and websites say, “Yes”
- Lots of physics articles say, “No”
 - Different reasons!

Answers: What Students Miss

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

Answers: What Students Miss

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- No. Evolution provides a pathway that maximizes the rate of entropy production!

Answers: What Students Miss

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- No. Evolution provides a pathway that maximizes the rate of entropy production!
- No. No step in the mechanism for evolution violates the Second Law, so the whole thing can't.

Answers: What Students Miss

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- No. Evolution provides a pathway that maximizes the rate of entropy production!
- No. No step in the mechanism for evolution violates the Second Law, so the whole thing can't.
- Not really. Entropy of the surroundings is so big, anything can happen.

Answers: Not Really?

*Does the Second Law of Thermodynamics contradict
the Theory of Evolution?*

Answers: Not Really?

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- One estimate...

Answers: Not Really?

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- One estimate...
- Entropy of biosphere *decreases* by 300 J/K per second.

Answers: Not Really?

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- One estimate...
- Entropy of biosphere *decreases* by 300 J/K per second.
- Entropy involved in our having a sun is 420×10^{12} J/K!

Answers: Not Really?

Does the Second Law of Thermodynamics contradict the Theory of Evolution?

- One estimate...
- Entropy of biosphere *decreases* by 300 J/K per second.
- Entropy involved in our having a sun is 420×10^{12} J/K!
- Is he saying evolution *does not* obey the Second Law, but is within the expected range of fluctuations?

Take-Home Messages

Take-Home Messages

- Faith-Integration assignments can be scientifically rigorous

Take-Home Messages

- Faith-Integration assignments can be scientifically rigorous
- You might learn something, too